Microelectronic Gas sensors for Non-invasive Analysis of Exhaled Gases

Aroutiounian VM*

Yerevan State University YSU, 1 Alex Manoukian Str, Armenia

ABSTRACT

Interest in the research of small-sized gas sensors (especially semiconductor) for medical applications has sharply increased in recent years. Requirements for to such sensors are their high sensitivity, performance and stability to very low concentrations of gases, mainly exhaled by a living organism. Today the doctor not only polls, and examines the patient but also sends it to the lab-tests of blood, urine, perform an electrocardiogram, etc. But, for example, today’s periodic monitoring of glucose concentrations in the blood induces acute pain, risk of virus (e.g., hepatitis B) infection from needles, and other difficulties in frequent monitoring. Even in 1784, Antoine Lavoisier wrote that of all the phenomena of life, none is more striking and worthier for physicists and physiologists than phenomena accompanying the breath. Meanwhile, one of the most important directions of modern medicine is non-invasive diagnostics of the patient, based on the analysis of exhaled air in a special device.

Keywords: Microelectronic semiconductor gas sensor; Breath analysis; Metal oxide

INTRODUCTION

We are talking about the exhaled air, its chemical composition, and volatile substances from the human body through the lungs. It is only necessary to make exhalation and after a few seconds or minutes to get the results of the analysis. The prospect of an non-invasive diagnosis is obvious - no unpleasant sensations, there is full safety from viral hepatitis, AIDS and other infections [1].

Exhaled breath contains about 1000 volatile organic compounds (VOCs) that are the products of metabolism. The exhaled gases such as NO₂, acetone, ethanol, NH₃, H₂S, and H₂O₂ can be used as sensors to diagnose various diseases. Several important gases having immediately dangerous to life or health concentration in air. For example, the threshold limit value for ethanol and acetone are 1000 and 750 ppm, correspondingly. Depending on the level of the basic exchange, a person releases through the lungs on average about 5 to 18 liters of carbon monoxide CO and 50 grams of water per hour. And with them - a huge number of micro-impurities volatile compounds. That is, a multi-component gas mixture formed, which accumulates first in the alveolus of the lungs. It is necessary to look for the cause of this either in incoming food (sources of volatile substances), or in internal organs, blood, blood vessels, sweat or urine.

According to the results of the study of volatile substances exhaled by a person, it is possible to judge the nature of nutrition - sufficiency of carbohydrates, excess fat and alcohol. By excess or lack of any chemical component in the spectrum of exhaled air, hereditary enzymopathy and the presence of various diseases can also be assumed. Due to the large surface of the lungs, volatile substances (ethanol, ammonia, acetone, and others) very quickly pass from the bloodstream to the external environment with exhaled air. It is the mixture of various molecules secreted by man that makes up the individual, unique smell of the patient. In a number of cases, this smell allows to immediately make the correct diagnosis. The sweet “liver odor” of explosives is due to a violation of the exchange of aromatic compounds and the accumulation of the product of the conversion of methionine - methyl mercaptan. This smell of raw liver often haunts pancreatic cancer patients and may be one of the first symptoms of this ailment. The smell of ammonia in explosives is typical for kidney and uremia diseases, the “mouse smell” - for patients with hereditary phenylketonuria, the smell of syrup is in violation of the metabolism of fatty acids and the accumulation of keto acids and amino acids in the blood and urine, the smell of acetone - in patients with diabetes mellitus. A sharp specific smell from the mouth is a symptom of a number of diseases of the oral cavity and stomach (stomatitis, periodontal disease, gastritis, peptic ulcer, and stomach cancer). In cardiopulmonary insufficiency, unpleasant sour smell of oxidized metabol products associated with incomplete combustion of proteins, fats, and carbohydrates in the liver often comes from patients. An unusual persistent smell is a formidable symptom of a growing tumor of the anterior brain. It turned out that with a stroke with an unfavorable outcome, much less acetone is released in patients than in healthy people. At the same time, diabetics, who also fell...
into a coma, exhale tens and hundreds of times more acetone than healthy ones. The content of acetone and ethanol is significantly different from healthy individuals (in patients with diabetes mellitus, cardiovascular disease, in children with bronchial asthma, diasthesia, in pregnant women with toxicosis of the first half of pregnancy). We also note that it is possible to successfully detect using semiconductor sensors odorless substances, such as carbon monoxide (carbon monoxide) or carbon dioxide.

Capabilities of detection of various diseases using the analysis of the breath are discussed below.

Data on the surveillance capabilities of diseases in the allocation of specific gases are collected below in Table 1 [1,2]. Naturally, assembled here material cannot be considered as exhaustive one. Note also that the analysis of the breath allows to detect (diagnose) cancer of stomach and ovaries, rheumatoid disease, acute myocardial infarction, gum disease, tooth decay, bacterial imbalance on the tongue, impacted wisdom teeth, dehydration, sleep apnea, gastritis, stomach ulcers duodenal ulcers and other diseases can be carried out.

It is known that a qualitative breakthrough in studying the composition of explosives was made only at the beginning of the 20th century, when mass spectrography (MS) (Thompson, 1912) and chromatography began to be used. The revival of chromatography is associated with the names of the English scientists Martin and Sing, who in 1941 developed the method of distribution chromatography, for which they were awarded the Nobel Prize in chemistry in 1952. From the middle of the 20th century to the present day, chromatography and mass spectrograph are among the most widely used analytical methods for studying explosives. About 1000 volatile metabolites were determined by these methods in explosives, many of which are used as markers of inflammation [1]. Their specificity and sensitivity for the diagnosis of many diseases are determined. In addition to chromatography and mass spectrograph, radioimmune and enzyme-linked immunosorbent assays, spectrophotometry, fluorometric and chemiluminescent methods are used, the protein matrix and tumor necrosis factor are studied, etc. In medicine, when studying gas exchange, gas analyzers are used to measure the concentration of carbon dioxide, oxygen, and nitrogen in the inhaled and the exhaled gas mixture.

<table>
<thead>
<tr>
<th>Gases</th>
<th>Exhalation and diagnostic significance</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nitrogen oxide (NO)</td>
<td>Respiratory diseases [asthma, chronic obstructive pulmonary disease (COPD), etc.] Rhinitis digestive diseases [inflammation in the stomach (gastritis, hepatitis, colitis), including infection Helicobacter pylori digestive cancer] Hyperthermia Heavy-tortured sepsis Uremia</td>
</tr>
<tr>
<td>Carbon monoxide (CO)</td>
<td>Anemia (hemolytic, sideroblastic cell) Carboxy hemoglobinemia in acute and chronic exposures Lasting stay at intakes Giperbilirubinemia of newborn oxidative stress hematoma Hemoglobinuria Preeclampsia of infection Thalassemia respiratory diseases (asthma, COPD), infections of the respiratory infection Inflammation of lung</td>
</tr>
<tr>
<td>Ammonia (NH₃)</td>
<td>Diseases of the kidneys and liver (renal insufficiency in nephritis, idiopathic hypertension, atherosclerosis of renal artery, toxicosis and nephropathy of pregnant, toxic defects of kidneys, paucity of the liver in jaundice, hepatitis, cirrhosis of the liver, toxic hepatitis) Acute and chronic radiation sickness The metabolism of monoxide in light Uremia</td>
</tr>
<tr>
<td>Hydrogen (H₂)</td>
<td>Diseases of bodies of digestion (digestive disorders of infants Gastrointestinal Disorders anaerobic bacteria in the large intestine Malabsorption hydrocarbons)</td>
</tr>
<tr>
<td>Hydrogen Peroxide H₂O₂</td>
<td>Respiratory diseases (asthma, chronic obstructive pulmonary disease (COPD), lung cancer, weakened respiratory lung function, etc.) The acute and chronic radiation sickness Diabetes</td>
</tr>
<tr>
<td>Methane</td>
<td>Gastrointestinal disorders (Malabsorption hydrocarbons) Colorectal cancer</td>
</tr>
<tr>
<td>CS₃ and pentane</td>
<td>Risk factor in coronary artery diseases, Schizophrenia</td>
</tr>
<tr>
<td>Ethylene (C₂H₄)</td>
<td>Oxygen stress, Lipid peroxidation internals at acute myocardial infarction Destruction caused by free radicals Uremia</td>
</tr>
<tr>
<td>Ethane (C₂H₆) and pentane</td>
<td>Lipid peroxidation in liver transplant peroxide Marker-assisted oxidation of lipids Schizophrenia COPD Interstitial lung disease Asthma Cystic fibrosis Heart failure Inflammatory bowel disease</td>
</tr>
<tr>
<td>Methanol</td>
<td>Diseases of the central nervous system Lung and breast cancer</td>
</tr>
<tr>
<td>Ethanol</td>
<td>Diabetes</td>
</tr>
<tr>
<td>Acetone C₃H₆O</td>
<td>Alcoholism, the function of the pancreas in acute pancreatitis and execute-destructive dietary and severe balance failure at lung cancer Diabetes Chronic liver disease</td>
</tr>
<tr>
<td>Isotopic modification</td>
<td>Infection with the bacterium Helicobacter pylori passage of food through the gastrointestinal tract the overgrowth of bacteria lactose Digestion pancreatic Dysfunction Malabsorption liver dysfunction, including cirrhosis the metabolism of bile glucose metabolism</td>
</tr>
<tr>
<td>Vapors of urine</td>
<td>Gastroenterology</td>
</tr>
<tr>
<td>Pentane and its Derivatives</td>
<td>Breath and lung cancer. Acute myocardial infarction Heart titrant rejection Rheumatic arthritis The exacerbation of asthma</td>
</tr>
<tr>
<td>Alkanes (hexane, etc.)</td>
<td>Lung cancer Tuberculosis</td>
</tr>
<tr>
<td>Dimethyl and carbonyl sulfides</td>
<td>Lung cancer Cystic fibrosis Intra- and extraoral halitosis Chronic liver disease Hypermethioninemia</td>
</tr>
<tr>
<td>Isoprene</td>
<td>Lung cancer Castraic ulcer Heart failure</td>
</tr>
<tr>
<td>Hydrogen sulfide</td>
<td>Intraoral halitosis</td>
</tr>
<tr>
<td>Heptane, xylenes, octane, ethylbenzenes, styrene, decane, toluene, benzene, gaseous aldehyde and formaldehyde</td>
<td>Lung cancer</td>
</tr>
</tbody>
</table>

Table 1: Gases and exhalation and diagnostic significance.
to study blood gases and to measure the concentration of the indicator gas in the artificially created mixture when determining a number of respiration parameters (instruments for measuring gas concentration in a gas mixture). In recent years, ionic and proton mass spectroscopic methods and gas chromatography have been used in medicine, but such equipment is expensive and bulky.

Of course, the possibility of physical-chemical studies of gas-phase are not limited to the analysis. Samples of internal abdominal gases can collect and examine from different sections of the bronchi with bronchoscopy, from the stomach with gastroscopy, from the colon with colonoscopy, from the bladder with cystoscopy, from the uterus during hysteroscopy. Today we are talking about the possibility of developing individual metabolic profile of the patient.

The use of semiconductor gas sensors for detecting the concentration of detectable gas released in patients (acetone in diabetes, for example) is widely discussed in the literature [3-5]. Determination of the concentration of chemical compounds in explosives using gas sensors will allow the diagnosis of the disease already in the early stages and will make it possible to control treatment. In addition to nitrogen, oxygen, and water vapor, explosives contain various gases in a sick person. By the concentration of acetone and the amount of current or voltage passing through the semiconductor sensor, for example, one can judge the degree of diabetes, acute heart failure, lung cancer, hydrogen peroxide and nitric oxide-asthma and other pulmonary diseases, cancer of the digestive system; ammonia and hydrogen sulfide - hepatis and cirrhosis; hydrogen and methane, a number of diseases of the digestive system, etc.

The current study of head, ovarian, bladder, prostate, kidney, gastric, and neck cancer, Crohn’s disease, ulcerative colitis, idiopathic and atypical Parkinson’s, multiple sclerosis, pulmonary hypertension and artificial intelligence methods.

Metal oxide semiconductors n-type SnO2, ZnO, WO3, TiO2, MoO3, In2O3, Fe2O3, and p-type CuO, NiO, Cr2O3, Mn3O4 are used during breath analysis [1,2,6-12, Haick et al. [13,14], Di Natale et al. [15], Righiottoni et al. [16], and Kim et al. [17]. Some interesting papers about breath analysis listed in Refs. [18-32]. Note that the disease diagnosis using exhaled breath is still in the nascent stage and needs further improvement for clinic applications.

DETECTION OF NO2 AND NH3 FOR THE DIAGNOSIS OF ASTHMA AND RENAL DISEASE

Vill-like WO3 nanostructures show high gas responses (Rgas/Rair) ~ 30 to 0.2 ppm NO at 200°C [33]. It is shown also in [34] that WO3 nanotubes exhibit high responses to 1 ppm NO. These results were obtained in highly humid (RH > 80%) atmospheres. BOSCH Healthcare Solutions announced a monitor which allows taking measurements of fractional exhaled nitric oxide (FENO) for the diagnosis of asthma [35]. Therefore, the potential of WO3 nanostructures for the diagnosis of asthma was demonstrated. This device consists of pretreatment components that precondition the exhaled gas and sensing components that measure FENO.

The initial stages of renal disease can be diagnosed by measuring NH3 vapor in exhaled breath [36]. End-stage renal disease (ESRD) requires time-consuming, expensive, and inconvenient hemodialysis, which significantly decreases the quality of life for patients. If the nephron function is completely lost, kidney transplantation needs to be considered. The breath NH3 concentration of ESRD patients ranges around 4.9 ppm in contrast to 0.96 ppm for healthy individuals [37]. Exhaled ammonia can be also used to diagnose liver disease [38] and helicobacter pylori infection [39].

β-MoO3 [40] and Si-doped MoO3 [41] have high selectively to NH3 detection at sub-ppm levels even in highly humid atmospheres, and can used for diagnosis of renal failure. MoO3, thin films [42-44], WO3, thin films with PtSiO2 overlayers [45], and Cr-doped WO3 films [46,47] have been reported to exhibit NH3-detection selectivity in a dry atmosphere. MoO3 and WO3 often show high response to basic amines such as trimethylamine, triethylamine, and butylamine [48-50].

DETECTION OF ACETONE FOR DIAGNOSING DIABETES

Diabetes patients use fat instead of glucose for energy and ketones are produced by the liver during fatty-acid metabolism. The concentrations of breath acetone in diabetes patients are reported to be higher than 1.8 ppm while those of healthy people are lower than 0.8 ppm [51]. Breath acetone concentrations can increase during ketogenic or low carbohydrate diets that are known to induce ketosis [52]. It allows checking the effectiveness of ketosis-based dietary programs for healthy people [53].

Among all other sensing materials, metal oxides show the highest responses to acetone. Various oxide sensing materials mentioned below allow to enhance selectivity and sensitivity to acetone under highly humid atmospheres.

There are Pt, Rh or Ni-loaded WO3, hemitubes/nanofibers, having an average size of 2 to 3 nm [54-57], Si-doped ε-WO3 [58-60], Pt-loaded SnO2 nanotubes/hierarchical nanofibers [61,62], Pd-loaded ZnO/ZnCo2O4 hollow spheres [63], RuO2-loaded WO3 nanofibers [64], Rh3O4-loaded WO3 nanofibers [65], and Pd-loaded Co3O4 hollow nanocages [66]. These fabricated nanostructures show great promise to be utilized as portable breath sensors for diabetes disease diagnosis.

Note that SnO2, ZnO, Fe2O3, and other metal oxide gas sensors often show similar responses to acetone and ethanol [67-69]. Accordingly, analyses from intoxicated diabetes patients, or the presence of small amounts of alcohol in the mouth, may not provide reliable diagnoses. Pure and catalyst loaded WO3 have been used as the most common and representative sensing materials for the selective detection of acetone [54-65].

DETECTION OF H2S FOR DIAGNOSING HALITOSIS

The degradation of S-containing amino acids in the oral cavity, upper/lower respiratory tract and alveolar exchange with blood lead to halitosis. It is detectable as highly odorous gases with concentrations less than 1 ppm, that can be used to diagnose fetor hepaticus and metabolic disorder.

CuO alone and as additive materials to SnO2, In2O3, and ZnO has been used as sensing material. The conversion of the p-type semiconductor CuO into metallic CuS leads to form junction n-type semiconductor-metallic CuS. The doping of hollow SnO2 spheres with CuO decreased the humidity dependence of the sensing characteristics to negligible levels without sacrificing high selectivity and sensitivity H2S at 80% RH. Therefore, CuO is effective both in dry air and exhaled breath.

Liang et al. [70] reported that the CuO-loaded In2O3 nanofiber sensor showed reversible H2S sensing characteristics above 300 °C.
Doping Mo on ZnO nanowires not only increased selectivity to H$_2$S but also significantly enhanced the reversibility of the H$_2$S sensing characteristics. Ag was also reported to be an effective additive that enhanced the H$_2$S selectivity of sensors based on SnO$_2$ [71,72], TiO$_2$ [73], and Fe$_2$O$_3$ [74]. Yoneda et al. [75] provided a mini review on various techniques for the analysis of halitosis.

**DETECTION OF VOLATILE ORGANIC COMPOUNDS FOR LUNG CANCER DIAGNOSIS**

Researchers have investigated different gases of lung cancer by comparing the breaths of healthy people and lung cancer patients [76-80]. Volatile organic compounds (VOCs) are reported as biomarker gases of lung cancer. Comprehensive review on VOCs related to lung cancer has been provided by Hakim et al. [81]. Note that most lung-cancer biomarker gases, except a few, contain benzene rings. Pure CNT or graphene-based sensing materials generally do not exhibit notable responses to large gases at room temperature or temperatures less than 100°C.

P-type semiconductors as gas sensing materials for benzene-derived gases, such as xylene, toluene, and benzene itself, can be used during measurements of lung cancer.

Gaseous aldehyde and formaldehyde breath biomarkers of lung cancer were proposed [82]. Ni-doped and Co-doped ZnO nanowires, Co$_3$O$_4$ (mesoporous, Cr-doped nanocomposites, and Pd-loaded shell sphere and hierarchical nanosheets), Cr$_2$O$_3$-ZnCr$_2$O$_4$ nanocomposites, NiONiMoO$_4$-SnO$_2$ nanocomposites, Cr-doped NiO hierarchical nanosheets, PdSnO$_2$ film with Co$_3$O$_4$ overlayer are very promising materials for detection of lung cancer. [3].

Of course, lung cancer can be diagnosed via the detection of a single biomarker gas. However, to increase diagnosis precision, the diagnosis of lung cancer using the electronic nose technology is preferable. Note also that pretreatment components such as dehumidifiers, preconcentrators, and flow sensors are very important for precise analysis. To miniaturize the system and achieve in situ diagnosis, the adsorption of analyte gases, interference gases, and moisture needs to be significantly improved [3].

**DETECTORS OF GASES**

An intensive work has been started on a manufacture of corresponding new-sized equipment using semiconductor gas sensors. Note that at Yerevan State University (Department of Semiconductor Physics and Microelectronics and the Scientific Center for Semiconductor Devices and Nanoelectronics) as a result of many years of research, semiconductor sensors of various gases have been developed. Sensors at YSU are sensitive to vapors of acetone, ammonia, nitrogen oxides, iso-butane, ammonia, hydrogen sulfide, hydrogen, various alcohols, toluene, combustible gases, hydrogen peroxide, propane, propylene glycol, formaldehyde, dimethylformamide, dichloroethane, gasoline, carbon monoxide and smoke arising in the early stages of a fire (see some references [83-98]). As part of a NATO grant, small-sized semiconductor sensors for nerve gases of chemical weapons (sarin, mustard gas) have been developed in YSU. Successful tests of such sensors were carried out at the Czech Military Academy.

Of course, there is information about commercial gas sensors (except for military, poisonous and a number of industrial gases) produced in the USA, Japan, China, Russia and Germany. Note that the sensors at YSU are stable in time, of smaller volume, much cheaper than produced in above-mentioned countries and are easily compatible with integrated circuits.

Figures 1-3 show our small-sized detectors of hydrogen and alcohols [83,99,100]. Figure 3 shows the possibility of implementing such a detector using a programmable board and the Arduino Nano memory. There are a couple of dozen contacts on the board, to which you can connect not only a number of chemical gas sensors developed by us but also all kinds of components: displays, LEDs, other sensors, motors, routers, magnetic locks, etc. It is possible to load a program into the Arduino processor that will control all of these devices according to a given algorithm. The Arduino board provides extensibility and the ability to automate certain activities. Thus, the realization of the electronic nose is possible, allowing the detection of several diseases. It is noteworthy that as a gas sensor can be used as sensors proposed in the center of semiconductor devices and nanotechnology of YSU, as well as any other, for example, Chinese gas sensors. In addition to the Arduino board...
and the gas sensor, we needed an alarm system, a servo drive that will simulate a valve, and a relay. We also added two buttons - the first one to turn the system on and off, and the second one to turn the servo on and off. The entire system is powered by six finger batteries or an AC-DS adapter. Earlier, we developed also a device, which allows us to measure simultaneously three gases – methane, carbon monoxide, and hydrogen.

Based on the sensors available at YSU, we are already developing testers for several diseases and gases for medical applications using explosives. Currently, work is underway to create devices with displays, indications of the degree of disease of which must be carefully established with the participation and help of medical doctors.

CONCLUSIONS

One of the most important directions of modern medicine is non-invasive diagnostics of the patient, based on the analysis of exhaled air in a special device. Data on the surveillance capabilities of various diseases in the allocation of specific gases are collected. The potential of various semiconductor chemical resistors made from metal oxide semiconductors to diagnose disease has been reviewed. For disease diagnosis from exhaled breath, pure and catalyst-loaded semiconductor chemical resistors sensing materials for specific biomarker gases are reported. In particular, nanosensors were discussed having the potential to detect asthma, renal disease, diabetes, halitosis, and lung cancer. The humidity and temperature dependencies of various sensing characteristics need to be decreased to negligible levels through the complete understanding of gas-sensing mechanisms and the interaction between moisture and the sensing surface. Corresponding low-cost detectors are developed which can detect small concentrations of exhaled air.

REFERENCES


68. Liang X, Kim TH, Yoon JW, Kwak CH, Lee JH. Ultrasensitive and ultrasensitive detection of H$_2$S using electrosprun CuO-loaded In$_2$O$_3$ nanofiber sensors assisted by pulse heating. Ibid. 2015;B209:934-942.


