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## Safety Assessment of Food Flavor - Cinnamaldehyde

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A flavor is a substance which may be a single chemical entity or a blend of chemicals of natural or synthetic origin whose primary purpose is to provide all or part of the particular flavor or effect to any food or other product taken orally. Food flavorings are an essential element in foods. Flavoring substances can be divided into artificial flavoring substances and nature-identical flavoring substances. Artificial flavoring substances are those substances which have not yet been noticed in natural products whereas nature identical flavoring substances are present in natural products intended for human consumption. Toxicological studies on flavoring compounds should be made according to necessity and adequacy. If a substance is consumed at high levels and there is no previous experience, with its use or it does not occur naturally in food, more direct toxicological data are essential [1,2]. The interpretation of toxicological tests requires the determination of a dosage level at which no adverse effects are observed. Aldehydes occur as natural (flavoring) constituents in a wide variety of foods and food components. They have been grouped into saturated aldehydes (e.g., formaldehyde, acetaldehyde), α, β- unsaturated aldehydes (e.g., acrolein and crotonaldehyde), and aldehydes containing a second functional group (e.g., furfural, cinnamaldehyde).

Cinnamaldehyde occurs naturally in Chinese cinnamon oil from the leaves and twigs of Cinnamomum cassia Blume. It has also been identified in Sri Lanka, Seychelles, and Japanese (Cinnamomum loureirii Nees) cinnamon bark and in other cinnamon species in varying amount. Thus, the genus *Cinnamomum* becomes the major source of cinnamaldehyde. We can notice various species of the genus Cinnamomum in Madagascar, Myanmar, Vietnam, Cambodia; Laos, Indonesia, Sumatra and Malaysia [3]. Cinnamaldehyde is a pale yellow liquid with a warm, sweet, spicy odor and pungent taste reminiscent of cinnamon. It has the following properties: mp-7.5°C; bp 252°C at 101 kPa (760 mm Hg), with partial decomposition, bp 128- 130°C at 2.7 kPa (20 mm Hg); specific gravity at 20°C: 1.1102; refractive index at 20°C: 1.61949. It is oxidized to cinnamic acid when exposed to air [4]. Cinnamaldehyde is widely used as a flavoring agent with a maximum permitted levels as high as 6400 ppm in fruits and juices, 3500 ppm in baked goods, 2200 ppm in breakfast cereals, 2000 ppm in baby food and desserts, and 1100 ppm in chewing gum. Cinnamaldehyde is also used as a fragrance in cosmetics, soaps and detergents. Cinnamaldehyde is often used as a stomachic, an antipyretic and an antiallergic drug and as a tonic in traditional Chinese medicines [5,6].

The acute toxicity of cinnamaldehyde is relatively low, with LD50 values ranging from 0.6 to more than 2 g/kg in various species. The LD50 of cinnamaldehyde in white rats, white mongrel mice and guinea pigs was reported as 3.4 g/kg. The LD50 of cinnamic aldehyde in mice by the ip route was reported as 2.318 g/kg. The acute dermal LD50 in rabbits was reported as 0.59 ml/kg by Shelanski [7-10]. Cinnamaldehyde in toothpaste could be the reason for allergic contact dermatitis and allergic contact stomatitis. North American Contact Dermatitis Research Group suggested that cinnamaldehyde might be a frequent cause of allergic reactions to perfumes [11-15].

Cinnamaldehyde is a widely used flavoring agent, and some 180,000 kg of it is consumed each year in foods, 39,000 kg from the use of cinnamon and 141,000 kg deliberately added as a flavor.

It has been estimated that 95% of its consumption is as a flavoring material and that its usage will grow by about 3% per year over the next 5 years [16,17]. Thus, cinnamaldehyde has a high potential for human consumption in the world. FEMA (The Flavor and Extract Manufacturers' Association of the USA) has given GRAS (Generally Recognized as Safe) status for cinnamaldehyde. Cinnamaldehyde is also approved for use by the Food and Drug Administration of the United States. The Joint FAO/WHO Expert Committee on Food Additives at first set an Acceptable Daily Intake (ADI) of 1.25 mg/ kg body weight. The World Health Organization has established a temporary ADI of 0.7 mg/kg body weight [18-20]. From our studies, we suggest that WHO should lower the suggested ADI level (0.7 mg/ kg body weight) for cinnamaldehyde [21-25]. The Council of Europe included cinnamaldehyde in the ADI of 1.25 mg/kg for total cinnamyl compounds. RIFM (Research Institute for Fragrance Materials) has observed its potential for sensitization and limited the use in perfumes for skin contact at 1% in the formula. Eugenol and limonene have been in conjunction with cinnamaldehyde as quenchers to neutralize the irritation reaction that some individuals have towards it [4,7,26]. In this context, regulatory agencies, researchers and policy makers may need to pay more attention to the possible toxicity of food flavorings.

## References

- Stofberg J (1988) Consumer exposure to chemically defined flavoring substances. PerfumFlavorist 13: 19-22.
- Oser BL, Hall RL (1977) Criteria employed by the expert panel of FEMA for the GRAS evaluation of flavouring substances. FoodCosmetToxicol 15: 457-466.
- Poole SK, Poole CF (1994) Thin-layer chromatographic method for the determination of the principal polar aromatic flavor compounds of the cinnamons of commerce. Analyst 119:113-120.
- Kirk-Othmer (1993) Cinnamic acid, cinnamaldehyde and cinnmyl alcohol. Kirk-Othmer Encyclopedia of Chemical Technology, Wiley, New York, pp. 344.
- Blakemore WM, Thompson HC (1983) Trace analysis of cinnamaldehyde in animal feed, human urine, and waste water by electron capture gas chromatography. J Agric Food Chem 31: 1047-1052.
- Katayama M, Mukai Y, Taniguchi H (1990) High-performance liquid chromatographic determination of cinnamaldehyde. Analyst 115: 9-11.
- Opdyke DLJ (1979) Monographs on fragrance raw materials. Food CosmetToxicol 17: 687-689.
- Zaitsev AN, Rakhmanina NL (1974) Some data on the toxic properties of derivatives of phenylethyl and cinnamic alcohol. Vopr Pitan 5: 48-53.
- Sporn A, Dinu I, Stanciu V (1965) Cercetareatoxicitatiialdehideicinamice. Igiena 14: 339.

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- 10. Shelanski MV(1973) Report to RIFM.
- 11. Drake TE, Maibach HI (1976) Allergic contact dermatitis and stomatitis caused by a cinnamic aldehyde-flavored toothpaste. Arch Dermatol 112: 202-203.
- Kirton V, Wilkinson DS (1973) Contact sensitivity to toothpaste. Br Med J 2: 115-116.
- Magnusson B, Wilkinson DS (1975) Cinnamic aldehyde in toothpaste. 1. Clinical aspects and patch tests. Contact Dermatitis 1: 70-76.
- 14. Millard LG (1973) Contact sensitivity to toothpaste. Br Med J 1: 676.
- 15. Schorr WF (1975) Allergic skin disease caused by cosmetics. Am Fam Phys 12: 90-95.
- Stofberg J, Grundschober F (1987) The consumption ratio and food predominance of flavoring materials. Perfum Flavorist 12: 27-56.
- 17. Clark GS (1991) Cinnamic aldehyde. Perfum Flavorist 16:25-30.
- FEMA (Flavor and Extract Manufacturers Association of the USA) (1965)
  Survey of flavoring ingredients usage levels No. 2286. Food Technology, Champaign, 19:155.
- 19. JECFA (FAO-WHO Joint Expert Committee on Food Additives) (1967) Specifications for the identity and purity of food addictives and their toxicological evaluation: some flavoring substances and non-nutritive sweetening agents. WHO, Geneva.

- WHO (World Health Organization) (1984) Evaluation of certain food additives.
  Technol Rep Ser 710: 1-44.
- 21. Gowder SJ, Devaraj H (2008) Food flavor cinnamaldehyde induced biochemical and histological changes in the kidney of male albino wistar rat. Environ Toxic Pharma 26: 68-74.
- 22. Gowder SJ, Devaraj H (2006) Effect of food flavor cinnamaldehyde on the antioxidant status of rat kidney. B Clin Pharmacol Toxicol 99: 379-382.
- 23. Gowder SJ, Devaraj H (2010) Food flavor cinnamaldehyde induced certain behavioral and biochemical parameters of male albino wistar rat. J Med Sci 3: 101-109.
- 24. Gowder S, Devaraj H (2010) A review on the nephrotoxicity of food flavor cinnamaldehyde. Cur Bio Com 6: 106-117.
- Gowder SJ (2013) Food additives risk factors for renal failure. J Social 3: e122.
- Council of Europe (1974) Natural Flavouring Substances, their Sources, and Added Artificial Flavouring Substances. Council of Europe, Strasbourg 102: 145