

Quantitative Analysis of Paper Chromatography

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DESCRIPTION

Paper chromatography is a technique widely used in the field of chemistry for separating and identifying different chemical compounds present in a mixture. The technique is based on the principle of differential migration of the components of the mixture in a mobile phase, which is usually a liquid, and a stationary phase, which is a strip of filter paper. Paper chromatography is a simple, rapid, and inexpensive technique that requires minimal equipment and expertise, making it a popular choice for qualitative and quantitative analysis in various fields, including biochemistry, pharmaceuticals, and forensics. The basic principle of paper chromatography is the partition of the mixture components between the stationary and mobile phases, which results in differential migration rates depending on the physical and chemical properties of the compounds. The stationary phase, which is the filter paper, contains small pores that can trap the components of the mixture based on their molecular size, shape, and polarity. The mobile phase, which is usually a solvent, carries the mixture components along the filter paper and interacts with them based on their solubility and polarity. As the mobile phase moves through the stationary phase, the different components of the mixture separate into distinct bands, forming a chromatogram.

One of the main advantages of paper chromatography is its simplicity and versatility. The technique requires only a few basic materials, such as filter paper, solvent, and a spotting device, which makes it easy to perform even in a non-laboratory setting. Moreover, paper chromatography can be used to separate a wide range of compounds, including amino acids, sugars, lipids, and pigments, making it a useful tool for both qualitative and quantitative analysis. Additionally, the technique is highly sensitive, and even small amounts of the sample can be analyzed. However, paper chromatography also has some limitations that need to be considered. One of the main challenges is the optimization of the experimental conditions, such as the choice of the stationary and mobile phases, the spotting technique, and the drying time, which can affect the quality and reproducibility of the results.

Furthermore, paper chromatography is not suitable for the separation of compounds with similar properties, as they tend to overlap and form broad bands in the chromatogram, making it difficult to identify and quantify them accurately.

Despite these limitations, paper chromatography remains a popular technique for separating and identifying compounds in various fields. The technique has several variations, including ascending, descending, and radial paper chromatography, which can be used depending on the sample type and the separation requirements. Additionally, paper chromatography can be combined with other analytical techniques, such as spectroscopy and mass spectrometry, to enhance the sensitivity and specificity of the analysis.

CONCLUSION

In conclusion, paper chromatography is a powerful technique that offers many advantages for the separation and identification of chemical compounds. The technique is simple, inexpensive, and versatile, making it a popular choice in various fields of chemistry. However, it also has some limitations that need to be considered, and the experimental conditions need to be optimized to achieve reliable and reproducible results. With its wide range of applications and potential for further development, paper chromatography will continue to play a vital role in chemical analysis and research in the future.

Correspondence to: Nastry Brignol, Department of Chemistry, University of Assiut, El Fateh, Assiut Governorate, Egypt, E-mail: Brignol6@gmail.com Received: 01-Mar-2023, Manuscript No. JCGST-23-23317; Editor assigned: 03-Mar-2023, PreQC No. JCGST-23-23317(PQ); Reviewed: 23-Mar-2023, QC No. JCGST-23-23317; Revised: 03-Apr-2023, Manuscript No. JCGST-23-23317 (R); Published: 12-Apr-2023, DOI: 10.35248/2157-7064.23.14.505 Citation: Brignol N (2023) Quantitative Analysis of Paper Chromatography. J Chromatogr Sep Tech. 14: 505.

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