

Discussion On The Effects Of Various Preparation Methods On The Work Performance Of AgCl/Al₂O₃ Catalyst Cast Off For Diesel Exhaust Treatment

Jack Denur*

Department of Physics, Electric and Gas Technology, University of North Texas, USA

INTRODUCTION

Catalytic purification of diesel exhaust pollutants is among the most difficult and widely discussed topics in the field of environmental catalysts worldwide. With advancement of the economic system, diesel engines have greatly increased in recent years due to their low fuel usage, high heat effectiveness, and other benefits. However, the emission levels of diesel soot (PM) and nitrogen oxides (NO_x) stored in diesel engine combustion are quite high, particularly the emissions of diesel soot, which is damaging to the environment and human body. As a result, with the government's increased focus on environmental protection, the exhaust emissions of diesel vehicles have increased. Meanwhile, as global trade and shipping expand, the naval diesel engine premised on petroleum products pollutes the ocean ever more severely. As according statistics, the global fuel usage of marine engines is determined by calculating at 100 million tonnes per year, resulting in NO_x air pollution every year. Ships account for 40 to half of total NO_x emissions in advanced shipping nations (such as Norway). Emission levels from ships may be the primary source of pollution in regions with high-density ports and routes. One of the major factors influencing worldwide air quality is diesel exhaust emissions. The impregnation technique was used to produce an AgCl/Al₂O₃ catalyst to eliminate NO_x from diesel exhaust gas in order to achieve the objectives of purifying exhaust gas. The AgCl/Al₂O₃ catalyst is a type of HC-SCR (hydrocarbons selective catalytic reduction) catalyst. As reducing agents, the HC-SCR catalyst uses oxidized hydrocarbons and their O₂ derivatives to convert dangerous NO_x in exhaust system into N₂. This type of energy is an Ag form catalyst, and it is worth examining for its simple preparation method, relatively cheap, and great catalytic properties. In this analysis, the catalysts were formulated under four diverse circumstances by varying the order of NH₄Cl, AgNO₃, and -Al₂O₃ additions, and also the drying and sintering times, and indeed the work place of the catalyst in

diesel exhaust gas was meant to simulate by incorporating the gas. After feeding gas into another catalyst-equipped device and heating it, the relationship between both the NO_x conversion rate and temperature has been determined. Noble metals are primarily used as catalysts in the removal of NO_x from diesel exhaust. Noble metal catalysts, including such Ag/Al₂O₃, have high SCR (Selective Catalytic Reduction) exposure to low temperatures, but they have some drawbacks, including low N₂ selectivity, a limited temperature coefficient, and a high value. Because of these issues, noble metal catalysts are not widely used in diesel exhaust treatment. The chemical properties of a diesel engine are complex, and also the composition of exhaust gas differs widely depending on engine operating conditions. Diesel engine pollutants include diesel particulate matter (PM), hydrocarbons (HC), CO, and NO_x, among others. Because of the high oxygen content throughout the exhaust, the NO_x emission is higher than those of a gasoline engine. PM emissions are hundreds of times higher than gasoline emissions. As a result, reducing NO_x and PM emissions is the primary focus of catalytic purification of diesel engine exhaust. The most viable NO_x removal technology is HC-SCR, which uses HC compounds (such as unburned HC in the exhaust or diesel fuel) as a reductant to decrease NO_x into N₂ in the exhaust of a diesel engine underneath the activity of a catalyst. The finding of this new reaction debunks the notion that NH₃ has been the only reductant capable of selectively reducing NO_x. The catalyst produced by conditions C and D has a high NO_x conversion rate over a wide range of temperatures. Simultaneously, the catalyst under condition D has high reliability after repetitive heating and air conditioning studies, so this catalyst is thought to have promising application prospects. A novel AgCl/Al₂O₃ catalyst was developed and tested. The -Al₂O₃ with a large surface area was loaded with AgCl. AgCl is being used as an active ingredient to decrease NO_x to N₂ in diesel exhaust through using hydrocarbons or oxygenated derivatives of hydrocarbons as a reduction agent, thus further

Correspondence to: Jack Denur, Department of Physics, Electric and Gas Technology, University of North Texas, USA, E-mail: jackD24@tx.edu

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lowering exhaust damage to the environment. Oxygen-containing compounds such as acetone (CH_3COCH_3), 2-propanol ($(\text{CH}_3)_2\text{CHOH}$), and ethanol ($\text{C}_2\text{H}_5\text{OH}$) are known to have better NO_x extraction efficiency than hydrocarbons. The variation among both oxygenated compounds and hydrocarbons in the reduction reaction is that if a catalyst uses hydrocarbons as a reduction agent, the percentage removal of NO_x is decreased in the presence of moisture vapour, whereas oxygenated derivatives of hydrocarbons can still maintain a good NO_x reduction effect in the same scenario. The main objective of this novel catalyst is to

lower the cost of catalytic reaction from a material standpoint. It will have a better NO_x reduction effect, and it has a high value in terms of popularization and use.

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DECLARATION OF CONFLICTING INTERESTS

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