

# Past and Present Research Systems of Green Chemistry

September 14-16, 2015 Orlando, USA

## SnO<sub>2</sub>-PbS nanocomposites and hetero structures: Fabrication, structures and applications

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SnO<sub>2</sub> is a significant metal-oxide, n-type wide band gap (3.6 eV at 300K) semiconductor. Due to its outstanding electrical, optical and electrochemical properties, SnO<sub>2</sub> offers a broad range of applications in solar cells, catalytic support materials, transparent electrodes and solid state chemical sensors. However, in terms of several of these applications, pure SnO<sub>2</sub> nanocrystals show a few classical drawbacks, mainly due to their large surface-to-volume ratio. Two standard strategies to advance the application-based performance of SnO<sub>2</sub> are its mixing and coating with another semiconductor material. Nowadays, nanocomposite and heterostructure materials comprised of two semiconductors of different band gap are proving of immense interest to researchers due to their superior electrical, optical, and electrochemical properties. In my presentation, I will discuss the fabrication, structures and application-based properties of SnO<sub>2</sub>-PbS nanocomposite and heterostructure materials. First I will describe some innovative solution and microwave synthetic techniques for the preparation of pure SnO<sub>2</sub> nanoparticles/nanospheres, pure PbS nanocubes, SnO<sub>2</sub>-PbS nanocomposites and heterostructures. The structures are confirmed by X-ray diffraction (XRD)/Reitveld study, Raman spectroscopy and transmission electron microscopy (TEM) analysis. Elemental mapping and line scan EDX analysis successfully demonstrate the effective PbS deposition on SnO<sub>2</sub> nanospheres as well as the mixing of SnO<sub>2</sub> and PbS to form nanocomposites. I will present a model to explain different relaxation processes in SnO<sub>2</sub>-PbS nanocomposite and heterostructure which helps to understand how the two semiconductors interact. Lastly, I will discuss some application-based properties of as-synthesized nanocomposite and heterostructure materials which are found to be significantly improved relative to pure SnO<sub>2</sub> or PbS.

### Biography

Arik Kar obtained his BSc (Hons) in Chemistry from Serampore College, University of Calcutta, in 2005. In 2007, he obtained his MSc degree in Chemistry from the University of Calcutta with a specialization in Inorganic Chemistry. He then completed his PhD in Materials Science from Jadavpur University in 2013. He was awarded a Newton International Fellowship from the Royal Society for the academic years 2014–2016. He is now continuing his research as a Newton International Fellow at the University of Cambridge. He has published 23 papers in reputed journals and has presented his research at 9 international conferences so far.

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