

**CALCIUM SODIUM PHOSPHOSILICATE : A PROMISING DESENSITIZING AGENT**<sup>1</sup> Ramoji Rao M.V.  
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**ABSTRACT**

Dentinal hypersensitivity is a common and chronic condition affecting the teeth in a stable percentage of population. Success with treatment of this condition has been limited at best. A new dentifrice containing Calcium Sodium Phosphosilicate material has been developed that results in a significant amount of tubule occlusion reducing tooth hypersensitivity. In this article, Calcium Sodium Phosphosilicate containing dentifrice, which is showing promise is analyzed, based on evidences.

**KEY WORDS :** Hypersensitivity, Dentinal tubule, Calcium sodium Phosphosilicate

**INTRODUCTION**

Dentine Hypersensitivity is short, sharp pain arising from exposed dentine. It occurs typically in response to chemical, thermal or osmotic stimuli and cannot be explained arising from any other dental defects or pathology<sup>1</sup>. Hypersensitivity is a common problem and estimates vary widely as to its prevalence. Some reports estimate that approximately 20% to 40% of adults suffer from dentine hypersensitivity. The incidence of tooth hypersensitivity increases with age and is attributed to the general increase in exposed root surfaces of the teeth from periodontal diseases, toothbrush abrasion, or cyclic loading fatigue of the thin enamel near the cemento-enamel junction<sup>2-4</sup>.

**Mechanism of Hypersensitivity**

The currently accepted theory for tooth hypersensitivity is the hydrodynamic theory proposed by Brannstrom<sup>5-6</sup>. This theory was based on the belief that open dentinal tubules allow fluid flow through the tubules, which excites the nerve endings in the dental pulp. Clinical replicas of sensitive teeth viewed under a Scanning Electron Microscope (SEM) reveal varying number of open or partially occluded dentinal tubules<sup>7</sup>. These studies have also shown that in patients with dentin hypersensitivity, there are a greater number of tubules per area and the diameter of the tubules is greater than in patients with no sensitivity<sup>8-9</sup>.

**Home Management of Hypersensitivity**

There are two basic approaches to the treatment and prevention of dentinal hypersensitivity. The first approach is to treat the tooth with a chemical agent that penetrates into the dentinal tubules and depolarizes the nerve synapse, which reduces sensitivity by preventing the conduction of pain impulses (e.g. potassium nitrate)<sup>10-11</sup>. The second approach is to treat the tooth with a chemical or physical agent (e.g., potassium oxalate, ferric oxalate, strontium chloride) that creates a deposition layer and mechanically occludes dentinal tubules, reducing sensitivity by prevention of pulpal fluid flow<sup>12-14</sup>. Although both approaches are effective at reducing or eliminating hypersensitivity, the duration of relief is highly variable. Hypersensitivity usually reappears due to toothbrush abrasion, the presence of acid challenges in the mouth, and/or degradation of the coating material<sup>12,15-20</sup>. Therefore, there is a need in the dental field for a material that chemically reacts with the surface of dentin, intimately adhere to tooth structure and significantly reduce the possibility of reopening dentinal tubules.

**Calcium Sodium Phosphosilicate**

Calcium Sodium Phosphosilicate is a bio-active glass in the class of highly biocompatible materials that were originally developed as bone

regenerative materials<sup>21</sup>. These materials are reactive when exposed to body fluids, and deposit hydroxycarbonate apatite, a mineral that is chemically similar to the mineral in enamel and dentin<sup>21-22</sup>. When incorporated into a dentifrice, particles are deposited onto the dentin surface to mechanically occlude dentinal tubules.

### Mode of Action of Calcium Sodium Phosphosilicate

The physical occlusion of Calcium Sodium Phosphosilicate particles begins when the material is subjected to an aqueous environment. Sodium ions ( $\text{Na}^+$ ) in the particles immediately begin to exchange with hydrogen cations ( $\text{H}^+$  or  $\text{H}_3\text{O}^+$ )<sup>21</sup>. This rapid release of ions allows Calcium ( $\text{Ca}^{2+}$ ) ions in the particles structure, as well as Phosphate ( $\text{PO}_4^{3-}$ ) ions to be released from the material. This initial series of reactions occur within seconds of exposure, and the release of the calcium and phosphate ions continues as long as the particles are exposed to the aqueous environment. A localized, transient increase in pH occurs during the initial exposure of the material due to the release of sodium. This increase in pH helps to precipitate the calcium and phosphate ions from the Calcium Sodium Phosphosilicate particle, along with calcium and phosphorus found in saliva, to form a calcium phosphate (Ca-P) layer. As the particle reactions continues and the deposition of calcium and phosphorus complexes continue, this layer crystallizes into hydroxycarbonate apatite which is chemically and structurally equivalent to biological apatite<sup>23</sup>. The combination of the residual Calcium Sodium Phosphosilicate particles and the hydroxycarbonate apatite layer results in the physical occlusion of dentinal tubules, which relieves hypersensitivity.

### Calcium Sodium Phosphosilicate - Evidences from Studies

Calcium Sodium Phosphosilicate products have received approval from the Food and Drug Administration. Numerous studies have focused on the decrease of sensitivity following the occlusion of open dentinal tubules. Most of these studies have used Scanning Electron Microscope to show the patent tubules before treatment and the occluded tubules after treatment<sup>2</sup>. A clinical study from China,<sup>24</sup> comparing 5% Calcium Sodium

Phosphosilicate - containing toothpaste to a Strontium Chloride positive control, showed that Calcium Sodium Phosphosilicate - containing toothpaste performs as well as or better than the positive control with respect to rapid relief of tooth hypersensitivity after two weeks and six weeks of daily use<sup>24</sup>.

Leonard Litkowski<sup>25</sup> evaluated bioglass as dentifrice ingredient capable of reducing tooth hypersensitivity through a mechanism of tubule occlusion. In his clinical study with cold air and tactile measures, showed significant reductions from baseline for test and control groups. The formulation with 7.5% Calcium Sodium Phosphosilicate significantly outperformed the control<sup>25</sup>.

In 2006, Burwell compared the use of DenShield, a dentifrice consisting of Calcium Sodium Phosphosilicate, with GC Tooth Mousse (containing Recaladent – caseinophosphopeptides-colloidal amorphous calcium phosphate) on bovine root surfaces in a pH-cycling model. After 10 days, the number of occluded dentinal tubules (as observed on SEM) was greater with DenShield than with the Tooth Mousse<sup>26</sup>.

### CONCLUSION

Current information supports blocking of dentinal tubules (either long term or short term) results in relief of sensitivity. This is in concurrence with Brannstrom's theory of hydrodynamic fluid movement in the tubules causing sensitivity (Brannstrom, 1963). All studies with Calcium Sodium Phosphosilicate containing dentifrices consistently showed that they are capable of significantly reducing hypersensitivity. Though these results are promising, further clinical trials are necessary to prove that Calcium Sodium Phosphosilicate is an ideal desensitizing agent.

### References

1. Addy M, Urquhart E. Dentine hypersensitivity : its prevalence, aetiology and clinical management. Dent Update 1992;19:407-412.
2. Gillam DG, SeoHS, Bulman JS, Newman HN, Perceptions of dentine hypersensitivity in a general practice population, J Oral Rehabil, 1999; 26(9):710-4.

3. Röss JS. The prevalence of dentine hypersensitivity in general dental practice in the UK. *J Clin Periodontol.* 2000 ; 27 (11): 860-5.
4. Fischer C, Fischer RG, Wennberg A. Prevalence and distribution of cervical dentine hypersensitivity in a population in Rio de Janeiro, Brazil. *J Dent* 1992; 20:272-6.
5. Brannstrom M, Astrom A. The hydrodynamics of the dentine; its possible relationship to dentinal pain. *Int Dent J.* 1972 ; 22(2):219-27.
6. Brannstrom M. The hydrodynamic theory of dentinal pain: sensation in preparations, caries, and the dentinal crack syndrome. *J Endod.* 1986; 12(10):453-7.
7. Arends J, Stokroos I, Jongebloed WG, Ruben J. The diameter of dentinal tubules in human coronal dentine after demineralization and air drying. A combined light microscopy and SEM study. *Caries Res.* 1995; 29(2):118-21.
8. Mordan NJ, Barber PM, Gillam DG. The dentine disc. A review of its applicability as a model for the in vitro testing of dentine hypersensitivity. *J Oral Rehabil.* 1997 ; 24 (2): 148-56.
9. Rimondini L, Baroni C, Carrassi A. Ultrastructure of hypersensitive and non – sensitive dentine. A study on replica models. *J Clin Periodontol.* 1995; 22(12):899-902.
10. Markowitz K, Kim S. The role of selected cations in the desensitization of intradental neves. *Proc Finn Dent Soc.* 1992; 88 Suppl 1; 39-54.
11. Schiff T, Dos Santos M, Laffi S, Yoshioka M, Brasil KD, Mccool JJ, De Vizio W. Efficacy of a toothpaste containing 5 % potassium nitrate and 1500 PPM sodium monofluorophosphate in a precipitated calcium carbonate base on dentinal hypersensitivity. *J Clin Dent.* 1998;9(1):22-5.
12. Orchardson R, Gillam D. The efficacy of potassium salts as agents for treating dentin hypersensitivity. *J Orofac Pain* 2000; 14 (1): 9-19.
13. Draglich WE, Pashley DH, Brennan WA, O'Neal RB, Horner JA, Van Dyke TE. An in vitro study of dentinal tubule occlusion by ferric oxalate. *J Periodontol.* 1993 ; 64(11):1045-51.
14. Dolci G, Mongiorgi R, Prati C, Valdre G (Calcium phosphates produced by physical methods in the treatment of dentin hypersensitivity.) *Minerva Stomatol.* 1999 ; 48 (10):463-76.
15. Kaufman HW, Wolff MS, Winston AE, Triol CW. Clinical evaluation of the effect of a remineralizing toothpaste on dentinal sensitivity. *J Clin Dent.* 1999; 10 (1Sec No):50-4.
16. Schiff T, Dotson M, Cohen S, De Vizio W, McCool J, Volpe A. Efficacy of a toothpaste containing potassium nitrate, soluble pyrophosphate, PVM/MA copolymer, and sodium fluoride on dentinal hypersensitivity : a twelve-week clinical study. *J Clin Dent.* 1994 ;5 Spec No :87-92.
17. Silverman G, Gingold J, Curro FA. Desensitizing effect of a potassium chloride toothpaste. *Am J Dent.* 1994;7(1):9-12.
18. Suge T, Ishikawa K, Kawasaki A, Yoshiyama M, Asaoka K, Ebisu S. Effects of fluoride on the calcium phosphate<sup>4</sup> precipitation method for dentinal tubule occlusion. *J Dent Res.* 1995; 74(4):1079-85.
19. West NX, Addy M, Jackson RJ, Ridge DB. Dentine hypersensitivity and the placebo response. A comparison of the effect of strontium acetate, potassium nitrate and fluoride toothpastes. *J Clin Periodontol.* 1997 ;234(4):209-15.
20. Yates R, Owens J, Jackson R, Newcombe RG, Addy M. A split – mouth placebo controlled study to determine the effect of amorphous calcium phosphate in the treatment of dentine hypersensitivity. *J Clin Periodontol.* 1998; 25(8):687-92.
21. Hench LL, Andersson O. Bioactive glasses. In: Hench LL, Wilson J, eds. *An Introduction to Bioceramics.* Vol 1. Singapore : World Scientific; 1993 : 45-47.
22. Andersson OH, Kangasniemi I. Calcium phosphate formation at the surface of bioactive glass in vitro. *J. Biomed. Mater. Res.* 1991; 25:1019-1030.
23. Litkowski LJ, Hack GD, Sheaffer HB, Greenspan DC. Occlusion of dentin tubules by 45S5 Bioglass®. In, *Bioceramics 10, Proceedings of the 10<sup>th</sup> International Symposium on Ceramics in Medicine, Paris, France, Oct. 1997.* eds. Sedel L, Rey C.
24. Du MQ, Tai BJ, Jiang H, Zhong JP, Greenspan DC, Clark AE. Efficacy of dentifrice containing bioactive glass (NovaMin ®) on dentine hypersensitivity. *J Dent Res, Spec Issue A* 2003; 82:1546.
25. Leonard Litkowski, DDS MS, Pilot clinical and in-vitro studies evaluating NovaMin ® in desensitizing dentifrice. *J Dent Res;* 1998 77: 1899-203.
26. Burwell A (2006). Tubule occlusion of a NovaMin ® - containing dentifrice compared to Recaldent – containing dentifrice – a remin / demin study in vitro. NovaMin Research Reports <http://www.novamin.com> (accessed 2-26-09)

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